

LETTERS

Effect of natural iron fertilization on carbon sequestration in the Southern Ocean

Stéphane Blain¹, Bernard Quéguiner¹, Leanne Armand¹, Sauveur Belviso², Bruno Bombled², Laurent Bopp², Andrew Bowie^{3,4}, Christian Brunet⁵, Corina Brussaard⁶, François Carlotti¹, Urania Christaki⁷, Antoine Corbière⁵, Isabelle Durand⁸, Frederike Ebersbach³, Jean-Luc Fuda⁹, Nicole Garcia¹, Loes Gerringa⁶, Brian Griffiths¹⁰, Catherine Guigue¹¹, Christophe Guillerme¹², Stéphanie Jacquet¹³, Catherine Jeandel¹⁴, Patrick Laan⁶, Dominique Lefèvre¹¹, Claire Lo Monaco⁵, Andrea Malits¹⁵, Julie Mosseri¹, Ingrid Obernosterer¹⁶, Young-Hyang Park⁸, Marc Picheral¹⁵, Philippe Pondaven¹⁷, Thomas Remenyi³, Valérie Sandroni¹, Géraldine Sarthou¹⁷, Nicolas Savoye^{13,18}, Lionel Scouarnec¹², Marc Souhaut¹⁴, Doris Thuiller⁵, Klaas Timmermans⁶, Thomas Trull^{3,10}, Julia Uitz¹⁵, Pieter van Beek¹⁴, Marcel Veldhuis⁶, Dorothee Vincent⁷, Eric Viollier¹⁹, Lilita Vong¹ & Thibaut Wagener¹⁵

The availability of iron limits primary productivity and the associated uptake of carbon over large areas of the ocean. Iron thus plays an important role in the carbon cycle, and changes in its supply to the surface ocean may have had a significant effect on atmospheric carbon dioxide concentrations over glacial–interglacial cycles^{1–5}. To date, the role of iron in carbon cycling has largely been assessed using short-term iron-addition experiments^{6,7}. It is difficult, however, to reliably assess the magnitude of carbon export to the ocean interior using such methods, and the short observational periods preclude extrapolation of the results to longer timescales⁸. Here we report observations of a phytoplankton bloom induced by natural iron fertilization—an approach that offers the opportunity to overcome some of the limitations of short-term experiments. We found that a large phytoplankton bloom over the Kerguelen plateau in the Southern Ocean was sustained by the supply of iron and major nutrients to surface waters from iron-rich deep water below. The efficiency of fertilization, defined as the ratio of the carbon export to the amount of iron supplied, was at least ten times higher than previous estimates from short-term blooms induced by iron-addition experiments⁷. This result sheds new light on the effect of long-term fertilization by iron and macronutrients on carbon sequestration, suggesting that changes in iron supply from below—as invoked in some palaeoclimatic^{9,10} and future climate change scenarios¹¹—may have a more significant effect on atmospheric carbon dioxide concentrations than previously thought.

The Southern Ocean plays a major role in the climate system, and is recognized as the oceanic body most sensitive to climate change^{12,13}. Iron fertilization of its surface waters during glacial times by enhanced dust deposition is a scenario (known as the ‘iron hypothesis’¹)

proposed to explain lower atmospheric CO₂ during colder climates. Different versions of this scenario^{2,3}, as well as the magnitude of the CO₂ drawdown induced by iron fertilization^{4,5}, are still debated, but the important role of iron in carbon cycling is no longer in question. The proximate control of biological productivity by iron in the Southern Ocean has been unequivocally demonstrated by iron addition experiments^{14–16}. All experiments show an enhancement of primary production associated with a shift in the structure of the phytoplankton community⁷. However, the short observational periods, as well as other intrinsic limits and artefacts of the small scale fertilization technique, have prevented a clear assessment of carbon export and preclude extrapolation to longer timescales.

Although the Southern Ocean is the largest high-nutrient low-chlorophyll (HNLC) region of the global ocean, natural phytoplankton blooms do occur in the vicinity of many Southern Ocean islands, as first reported in ref. 17 and subsequently characterized in detail by satellite images^{18,19}. The largest bloom is observed around Kerguelen Island and the adjacent plateau to its southeast (Fig. 1a, b). The bloom has two main features: (1) a narrow plume that extends northeast of the island and north of the Polar Front that shows high mesoscale and temporal variability²⁰, and (2) a larger bloom (~45,000 km²) southeast of the island and south of the Polar Front which is remarkably constrained to the bathymetry of the plateau. In 2004–05, this larger bloom began in early November, achieved high phytoplankton biomass (~3 µg chlorophyll *a* per litre) in December and January and then collapsed in late February (Fig. 1c).

The KEOPS (Kerguelen ocean and plateau compared study; 19 January to 13 February 2005) cruise carried out a survey inside and outside this bloom (Fig. 1b), with particular focus on two contrasting stations: A3 (50° 38' S, 72° 05' E) and C11 (51° 39' S, 78° 00' E).

¹Laboratoire d'Océanographie et de Biogéochimie, Centre Océanologique de Marseille, CNRS, Université de la Méditerranée, campus de Luminy, case 901, 13288 Marseille Cedex 09, France. ²IPSL/Laboratoire des Sciences du Climat et de l'Environnement, CEN de Saclay, Bât. 701 l'Orme des Merisiers, 91191 Gif-sur-Yvette, France. ³Antarctic Climate and Ecosystems CRC, Hobart, Tasmania 7001, Australia. ⁴ACROSS, School of Chemistry, University of Tasmania, Hobart, Tasmania 7001, Australia. ⁵LOCEAN-IPSL, UMR 7159, CNRS, Université P. et M. Curie, Case 100, 4 place Jussieu, 75252 Paris Cedex 5, France. ⁶Royal Netherlands Institute for Sea Research (NIOZ), PO Box 59, 1790 AB Den Burg, Texel, The Netherlands. ⁷FRE ELICO, Université du Littoral Côte d'Opale, Maison de la Recherche en Environnement Naturel (MREN), 32 avenue Foch, 62930 Wimereux, France. ⁸USM402/LOCEAN, Département des Milieux et Peuplements Marins, Muséum National d'Histoire Naturelle, 43 rue Cuvier, F-75231 Paris Cedex 05, France. ⁹Centre Océanologique de Marseille, Campus de Luminy, 13288 Marseille Cedex 09, France. ¹⁰CSIRO Division of Marine and Atmospheric Research, GPO Box 1538, Hobart, Tasmania 7001, Australia. ¹¹LMGEM UMR CNRS 617, Campus de Luminy, case 901, 13288 Marseille Cedex 09, France. ¹²DT INSU CNRS, Bât. IPEV BP 74, Technopole Brest Iroise, 29280 Plouzané, France. ¹³Department of Analytical and Environmental Chemistry, Vrije Universiteit Brussel, Pleinlaan 2, B-1050 Brussels, Belgium. ¹⁴LEGOS (CNRS/CNES/IRD/UPS), Observatoire Midi-Pyrénées, 14 avenue Edouard Belin, 31400 Toulouse, France. ¹⁵Laboratoire d'Océanographie de Villefranche, Quai de La Darse, BP 8, 06238 Villefranche-sur-Mer, France. ¹⁶Université Pierre et Marie Curie-Paris 6, UMR7621, CNRS, F66650 Banyuls-sur-Mer, France. ¹⁷UMR 6539/LEMAR/IUEM, Technopole Brest Iroise, Place Nicolas Copernic, 29280 Plouzané, France. ¹⁸Observatoire Aquitain des Sciences de l'Univers, UMR CNRS 5805 EPOC, Station Marine d'Arcachon, 2 rue du Pr. Jolyet, 33120 Arcachon, France. ¹⁹Laboratoire de Géochimie des Eaux, UMP IPGP 7154, Université Denis Diderot Paris 7, 2 place Jussieu, 75251 Paris Cedex 05, France.

Measurements of the partial pressure of CO_2 (p_{CO_2}) in surface waters indicate that the bloom was an important sink for CO_2 (Fig. 1d), with a mean p_{CO_2} drawdown of $58 \pm 11 \mu\text{atm}$, which was $\sim 2\text{--}3$ times higher than that observed following Southern Ocean iron addition experiments, probably reflecting the longer duration of the Kerguelen bloom. Silicic acid was almost depleted ($1\text{--}2 \mu\text{M}$), while nitrate concentrations remained relatively high ($23 \mu\text{M}$) (Supplementary Fig. 1).

Measurements of dissolved iron (DFe) revealed that concentrations in the surface mixed layer were low on and off the plateau ($0.090 \pm 0.034 \text{ nM}$; mean \pm s.d., $n = 49$) and typical of surface waters of the open Southern Ocean^{15,21}. But vertical profiles of DFe revealed a major difference between on- and off-plateau stations below 150 m (Fig. 2b). Above the plateau, DFe concentrations increased with depth and reached a mean maximum of $\sim 0.35 \text{ nM}$ (range $0.19\text{--}0.51 \text{ nM}$) at 500 m, close to the bottom. In contrast, off the plateau (bottom $> 1,500 \text{ m}$) mean DFe concentrations were $\sim 0.18 \text{ nM}$ at 600 m. Organic ligand concentrations (L) were sufficient to keep the supplied DFe soluble ($L/\text{DFe} > 2.4$ for all stations). The low concentrations of DFe measured in surface waters both on and off the plateau are known to be close to rate-limiting for many phytoplankton from bottle²² and mesoscale iron addition experiments^{14–16}. Thus it appears that phytoplankton production depletes the iron to rate-limiting levels both off and on the plateau, with the higher biomass over the plateau induced by higher iron supply from below.

Availability of iron in surface waters over the plateau appears to be enhanced by the greater winter stock and by a higher ongoing supply from increased vertical mixing and the steeper DFe gradient. Between 100 and 200 m, the vertical diffusivity ($K_z = (3.3 \pm 3.3) \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$, see Table 1) was substantially higher than the most recent estimate of

K_z in the open Southern Ocean ($(0.11 \pm 0.2) \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$)²³. This enhancement of vertical mixing was probably due to internal wave activity (Fig. 2c, d), which also appears to enhance mixing at the C11 site off the plateau ($K_z = (3.2 \pm 2.3) \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$, see Table 1). Combining the K_z estimates with the DFe gradients (Table 1) suggests a nearly eightfold higher DFe flux into surface waters on the plateau ($31 \text{ nmol m}^{-2} \text{ d}^{-1}$) than off the plateau ($4 \text{ nmol m}^{-2} \text{ d}^{-1}$). For comparison, DFe fluxes around $3 \text{ nmol m}^{-2} \text{ d}^{-1}$ were found at the SOIREE and FeCycle HNLC sites^{23,24}.

The net DFe requirements of phytoplankton ($208 \pm 77 \text{ nmol m}^{-2} \text{ d}^{-1}$)—defined as the difference between the total uptake and regeneration rates of DFe, and estimated from our ⁵⁵Fe experiments carried out at the beginning (19 January) and at the end (12 February) of the survey—are not balanced by the vertical supply of DFe ($31 \text{ nmol m}^{-2} \text{ d}^{-1}$). This suggests an additional supply from, for example, ongoing depletion of the winter surface stock or from dissolution of lithogenic particulate Fe ($\text{PFe}_{\text{litho}}$). The total supply cannot exceed the demand, because this would lead to elevated iron concentrations in surface waters that would be in contradiction with our observations. Between 19 January and 12 February, the steady depletion of a winter stock by 0.06 nM or the application of a daily dissolution rate of $\sim 2.5\%$ of the observed $\text{PFe}_{\text{litho}}$ at A3 on 24 January ($\sim 0.1 \text{ nM}$ based on particulate Al and a crustal Fe/Al ratio) is sufficient to balance the demand. The accurate determination of these fluxes is out of reach of the present analytical capabilities, but the values are within realistic ranges¹¹ and are upper limits because the additional supply is most probably a combination of both processes. Constraining the balance of the demand with supply suggests an additional DFe input of $177 \text{ nmol m}^{-2} \text{ d}^{-1}$ at A3 (Table 1). Including this possible supply suggests an excess of DFe supply over

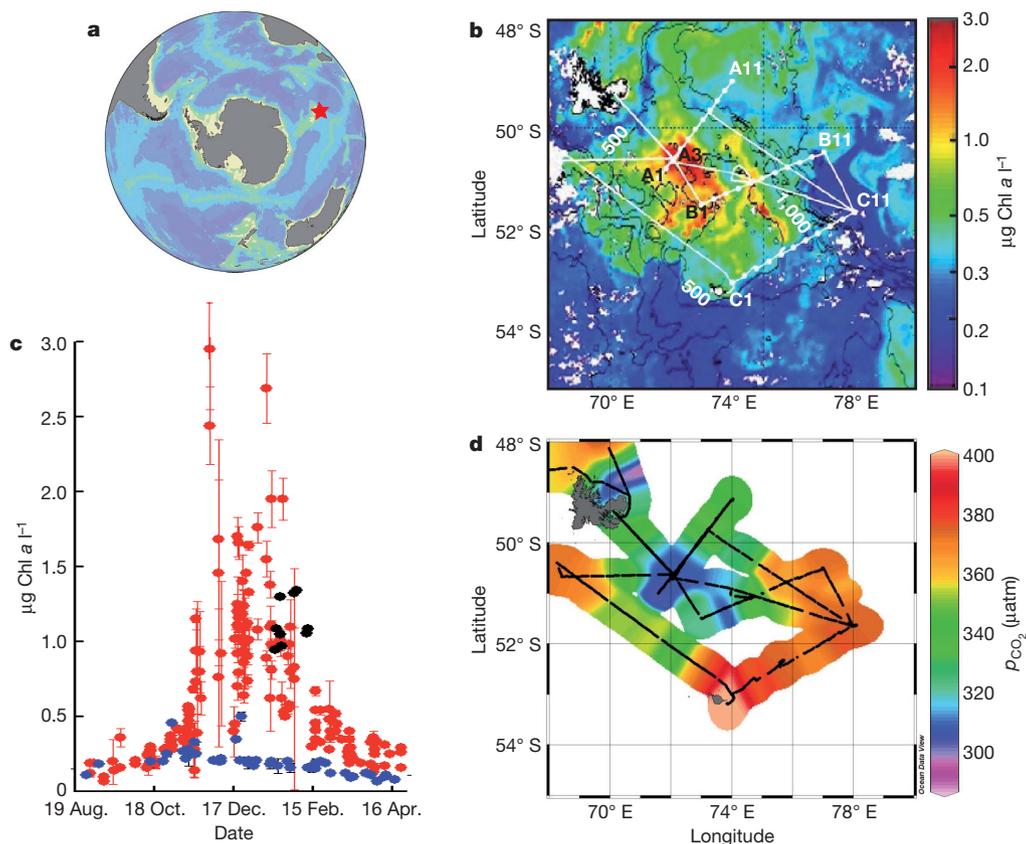


Figure 1 | Location of the study sites, temporal evolution of the bloom and surface water properties. **a**, Location of the Kerguelen plateau (red star) in the Southern Ocean. **b**, Satellite image of the bloom during the KEOPS cruise. The track of the cruise (white line), the position of the stations (white dots) and bathymetry (black lines) are shown. Chl *a*, chlorophyll *a*. **c**, Time series of satellite-determined Chl *a* at stations within the bloom (red dots)

and at C11 (blue dots). (MODIS results provided by CSIRO Marine Research, Hobart.) Error bars, ± 1 s.d. calculated from the individual passes of the satellite. Black dots denote high-performance liquid chromatography measurements in surface waters during the cruise. **d**, Surface p_{CO_2} measured by underway sampling.

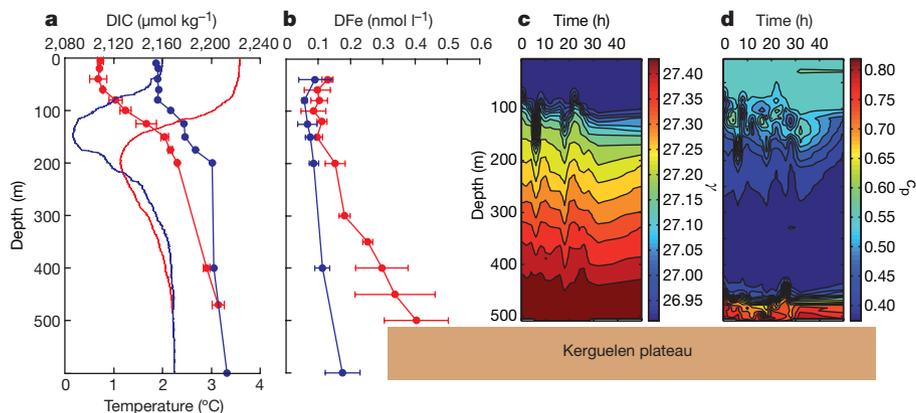


Figure 2 | Iron fertilization above the plateau. **a**, Mean profiles ($n = 3$) of DIC concentrations at A3 (red circles) and at C11 (blue circles). Red and blue lines denote typical temperature profiles at A3 and C11, respectively. Error bars, ± 1 s.d. **b**, Mean profiles ($n = 3$) of DFe concentrations at A3 (red circles) and outside the plateau (blue circles). Stations outside the plateau are C11, C9, B11, B9, A11 and A9. Error bars, ± 1 s.d. Above the Kerguelen plateau, close to the bottom, a large variability of the vertical DFe gradients

among the different profiles was observed, resulting in the large s.d. of the mean DFe value. **c**, Time series of vertical profiles of γ at station A3 ($\gamma = \text{water density} - 10^3$, where water density is in kg m^{-3}). **d**, Time series of vertical profiles of particle attenuation at 660 nm (c_p) at station A3. A cast was performed every two hours. The brown block labelled Kerguelen plateau represents the bottom at station A3.

the plateau, in comparison to surrounding waters, of $204 \pm 77 \text{ nmol m}^{-2} \text{ d}^{-1}$. This is a maximum estimate of the excess DFe supply, in that it assumes no additional supply to balance the demand at C11 as the demand was not determined at that location (Table 1).

Carbon dioxide uptake fuelled by the enhanced iron supply is clear from the lowered p_{CO_2} of surface waters (Fig. 1d), but how much carbon is exported from surface waters and sequestered at depth? We estimated carbon export using the deficit of ^{234}Th activity in the water column, and the ratio of particulate organic carbon (POC) to ^{234}Th (Fig. 3; Table 1).

The POC export indicated by the ^{234}Th technique in the bloom over the plateau was approximately twice as large as in the HNLC

waters off the plateau (Table 1), but still within the range of POC fluxes from the mixed layer estimated in the same way for blooms in the Polar Frontal Zone²⁵. At A3, the POC export flux represented about 30% of primary production. The mean excess of carbon export at A3 compared to C11 was 10.8 ± 4.9 and $14.2 \pm 7.9 \text{ mmol m}^{-2} \text{ d}^{-1}$ at 100 and 200 m, respectively. This can be compared to the results of SOFEX, the only Southern Ocean iron addition experiment that has measured enhanced POC export²⁶. The excess measured at 100 m was $7.1 \text{ mmol m}^{-2} \text{ d}^{-1}$, and extrapolated to 250 m was $3 \text{ mmol m}^{-2} \text{ d}^{-1}$.

On the basis of the excess of POC export at 200 m and the excess of DFe supply, we obtain a sequestration efficiency (that is, the excess of POC export divided by the excess of DFe supply) of $70,000 \pm$

Table 1 | DFe and carbon budget

Station	DFe		Carbon	
	Bloom (A3)	HNLC (C11)	Bloom (A3)	HNLC (C11)
Short-term fluxes*				
Vertical diffusivity ($10^{-4} \text{ m}^2 \text{ s}^{-1}$)†	3.2	2.4	3.0	3.8
Vertical gradient (mmol m^{-4})‡	11.2×10^{-7}	2.0×10^{-7}	0.56	0.32
Vertical supply ($\text{mmol m}^{-2} \text{ d}^{-1}$)§	3.1×10^{-5}	0.4×10^{-5}	14.6	10.6
Unaccounted supply ($\text{mmol m}^{-2} \text{ d}^{-1}$)	17.7×10^{-5}	ND	28¶	-2.7¶
Th-derived POC export ($\text{mmol m}^{-2} \text{ d}^{-1}$) at 100 m#			23.0	12.2
Th-derived POC export ($\text{mmol m}^{-2} \text{ d}^{-1}$) at 200 m#			24.5	10.3
Seasonal budget*				
Winter concentration (mmol m^{-3})⊗	1.53×10^{-4}	8.6×10^{-5}	2,230	2,237
Summer concentration (mmol m^{-3})**	8.6×10^{-5}	7.2×10^{-5}	2,167	2,214
Winter stock utilization (mmol m^{-2})††	4.7×10^{-3}	9.5×10^{-4}	4,410	1,564
Vertical supply (mmol m^{-2})‡‡	1.4×10^{-3}	1.9×10^{-4}	657	477
Air-sea supply (mmol m^{-2})‡‡			1,260	-121
POC accumulation in the ML (mmol m^{-2})§§			670	190
DOC accumulation in the ML (mmol m^{-2})§§			610	ND
POC export below the ML (mmol m^{-2})			5,047	1,730

HNLC, high-nutrient low-chlorophyll; ND, not determined; POC, particulate organic carbon; DOC, dissolved organic carbon; ML, mixed layer.

* See also Supplementary Table 1.

† Mean values corresponding to the depth stratum where the gradient is observed (150–200 m for DFe, and 80–150 m for DIC).

‡ Vertical gradients were determined using the linear part of the vertical profiles (Fig. 2a, b) corresponding to the depth stratum defined in the previous footnote.

§ Vertical supply was calculated by multiplying the vertical gradient with the mean vertical diffusivity coefficient converted to $\text{m}^2 \text{ d}^{-1}$.

|| This term is the supply of DFe needed to balance the mean net DFe demand ($208 \text{ nmol m}^{-2} \text{ d}^{-1}$) calculated as the difference between the total uptake rate ($423 \text{ nmol m}^{-2} \text{ d}^{-1}$) and the regeneration rate ($215 \text{ nmol m}^{-2} \text{ d}^{-1}$) measured during the cruise. This term was not determined at C11. See text for details.

¶ This term is the mean gas exchange flux calculated from two different parameterizations^{31,32} and the wind speed measured during the cruise.

See text and Methods for details.

⊗ Winter concentration (C_{winter}) was taken from the concentration measured at the depth of the temperature minimum characterizing the remnant winter water (Fig. 2a).

** Summer concentration (C_{summer}) was the mean value measured within the mixed layer during the KEOPS survey.

†† Seasonal apparent consumption was calculated using the equation ($C_{\text{winter}} - C_{\text{summer}}$)/MLD. The mean mixed layer depth (MLD) was 70 ± 20 m and 68 ± 13 m at A3 and C11, respectively.

‡‡ Vertical supply and air-sea supply were calculated assuming linear build-up of the gradients over 90 days.

§§ The accumulation of POC and DOC was the difference between the stock in summer and in winter. (no DOC data are available at C11).

||| POC export = DIC winter stock utilization + DIC vertical supply + DIC air-sea supply - POC accumulation - DOC accumulation. In a similar way to the POC export derived from Th fluxes, we assume that the carbon export at 200 m was roughly the same as just below the MLD.

46,000 mol mol⁻¹ above the Kerguelen plateau. To take into account the possible variability in the C export and DFe supply as the season progresses, we also calculated the sequestration efficiency of the natural fertilization based on the seasonal budgets (Table 1). This yields a much higher value of 668,000 mol mol⁻¹ (the ratios between the C export and the DFe supply were 9.9×10^5 and 1.5×10^6 at A3 and C11, respectively). It is evident from the individual terms of the budgets (Supplementary Table 1) that the uncertainty on the efficiency estimate is large, but we have investigated three possible biases, as follows. (1) The vertical diffusivity is difficult to constrain. However, the efficiency ratio is only weakly sensitive to its magnitude because the dissolved inorganic carbon (DIC)/DFe ratio of the resupply is similar to that of the export. (2) Our late-summer observations may under-estimate the iron supply, but even the extreme assumption that DFe concentrations over the plateau were much higher earlier in the year (for example, that the 500 m near-bottom DFe value of 0.40 nM at A3 characterized waters at 200 m throughout the rest of the year) still leads to a seasonal sequestration efficiency of 149,000 mol mol⁻¹. (3) The seasonal budget does not include an additional supply from the dissolution of PFe_{litho}. If we applied, over the whole season, the maximum PFe_{litho} dissolution flux determined in the short-term budget, this also leads to a high efficiency ratio of 159,000 mol mol⁻¹. The high efficiencies resulting from the short and seasonal budgets are consistent with our shipboard observations at A3 of C:Fe uptake rate ratios ($200,000 \pm 118,000$ mol mol⁻¹) and C:Fe contents of phytoplankton cells ($227,000 \pm 5,000$ mol mol⁻¹).

Our different approaches to estimating the sequestration efficiency give results that range over an order of magnitude, yet all the estimates are at least 10 times higher than the mean efficiency of $\sim 4,300$ mol mol⁻¹ estimated from SOFEX and other mesoscale iron addition experiments in the Southern Ocean⁷. Our estimates do not include horizontal transport of DIC and DFe, but mixing with waters outside the plateau would tend to supply DIC and remove DFe from the plateau and thus to increase the sequestration efficiency. Our ratio might be biased if a strong decoupling between carbon and iron cycling existed, but our measurements show that, in the bloom, similar percentages of the fixed C (40%) and Fe (50%) were

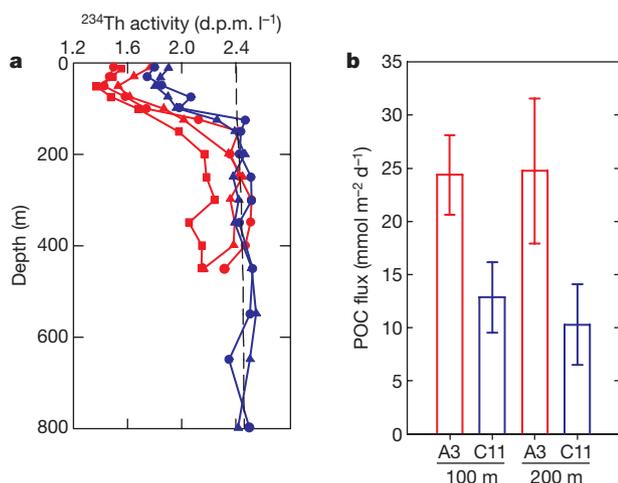


Figure 3 | Carbon export at A3 and C11. **a**, Profiles of ²³⁴Th activity at A3 (red lines) and C11 (blue lines). The date of the visits (in 2005) are as follows. At A3; 25 January (circles), 3 February (triangles), 12 February (squares); at C11; 26 January (circles), 05 February (triangles). The dashed line denotes the mean profile of ²³⁸U activity. **b**, POC fluxes at A3 and C11. The POC fluxes were derived from ²³⁴Th fluxes using a non-steady-state method and POC:²³⁴Th ratios (see Methods). The contribution of the vertical diffusivity to the fluxes was negligible at 200 m and around 10% at 100 m, at A3 and C11. The non-steady-state contribution was less than 4% at 100 m. At 200 m it was 34% and 29% at A3 and C11, respectively. Error bars, ± 1 s.d. of the total flux. At both depths, POC fluxes were substantially higher inside the bloom compared to outside.

remineralized as DIC and DFe, respectively. Thus the conclusion that natural iron fertilization over the Kerguelen plateau has a high carbon sequestration efficiency is quite robust.

The higher carbon sequestration efficiency of the natural bloom in comparison to mesoscale iron addition experiments derives from differences in both terms of the sequestration efficiency ratio. The excess of carbon export was higher during the KEOPS bloom than SOFEX, probably owing to its later stage of development (Supplementary Materials), and because during SOFEX the end of the bloom was not reached²⁶. The addition of DFe occurs slowly and continuously during natural fertilization, whereas purposeful additions of large amounts of iron within a short period lead to the loss of most (80–95%) of the added DFe during mesoscale enrichment experiments²⁷.

The occurrence of the Kerguelen bloom is contingent on persistent iron fertilization, but nevertheless its duration is due to the concomitant supply of macronutrients from surrounding waters and from below. The KEOPS results are tightly linked to the mode of iron supply, which is different from dust deposition or purposeful additions. Our estimate of carbon sequestration efficiency is important to the evaluation of palaeo hypotheses^{9,10} and future climate change scenarios¹¹ linking iron supply from below with ocean productivity and its influence on atmospheric carbon dioxide. The enrichment of deep water by terrigenous inputs¹⁰ or by the indirect effect of dust deposition⁹ has been proposed. The complex interplay between the iron and carbon cycles²⁸ precludes a direct extrapolation of how much CO₂ will be really removed from the atmosphere following the fertilization of the Southern Ocean from below, but clearly the natural system is extremely sensitive to iron, far more so than suggested by mesoscale iron addition experiments. However, for the reasons presented above, we emphasize that the high sequestration efficiency determined in the Kerguelen bloom should not be taken as an indication that controversial geoengineering CO₂ mitigation proposals^{29,30} will be able to obtain high efficiencies.

METHODS SUMMARY

Composite images from the satellite-borne MODIS (Moderate Resolution Imaging Spectroradiometer) and MERIS (Medium Resolution Imaging Spectrometer Instrument), processed by ACRI Co. and delivered daily to the RV *Marion Dufresne*, were used to locate the stations inside and outside the bloom. The vessel's seawater supply was used to monitor surface waters. Water column properties were measured *in situ* by instruments mounted on a rosette frame, and discrete samples were collected using Niskin bottles. *In situ* pumps were used to collect particles. All the samples were analysed applying standard protocols. Special attention was paid to Fe sampling and measurements, using trace metal clean techniques.

Full Methods and any associated references are available in the online version of the paper at www.nature.com/nature.

Received 22 March 2006; accepted 26 February 2007.

- Martin, J. H. Glacial-interglacial CO₂ change: The iron hypothesis. *Paleoceanography* **5**, 1–13 (1990).
- Brzezinsky, M. A. *et al.* A switch from Si(OH)₄ to NO₃⁻ depletion in the glacial Southern Ocean. *Geophys. Res. Lett.* **29**, doi:10.1029/2001GL014349 (2002).
- Sigman, D. M. & Boyle, E. A. Glacial/interglacial variations in atmospheric carbon dioxide. *Science* **407**, 859–869 (2000).
- Bopp, L., Kohfeld, K. E., Le Quéré, C. & Aumont, O. Dust impact on marine biota and atmospheric CO₂ during glacial periods. *Paleoceanography* **18**, doi:10.1029/2002PA000810 (2003).
- Watson, A. J., Bakker, D. C. E., Ridgwell, A. J., Boyd, P. W. & Law, C. Effect of iron supply on Southern Ocean CO₂ uptake and implications for glacial atmospheric CO₂. *Nature* **407**, 730–733 (2000).
- Boyd, P. W. *et al.* Mesoscale iron enrichment experiments 1993–2005: Synthesis and future directions. *Science* **315**, 612–617 (2007).
- De Baar, H. J. W. *et al.* Synthesis of iron fertilization experiments: from the iron age in the age of enlightenment. *J. Geophys. Res.* **110**, doi:10.1029/2004JC002601 (2005).
- Boyd, P. W., Jackson, G. A. & Waite, A. M. Are mesoscale perturbation experiments in polar waters prone to physical artefacts? Evidence from algal aggregation modelling studies. *Geophys. Res. Lett.* **20**, doi:10.1029/2001GL014210 (2002).

9. Ridgwell, A. J. & Watson, A. Feedback between aeolian dust, climate, and atmospheric CO₂ in glacial time. *Paleoceanogr.* **17**, 1059, doi:10.1029/2001PA000729 (2002).
10. Latimer, J. C. & Filipelli, G. M. Terrigenous input and paleoproductivity in the Southern Ocean. *Paleoceanography* **16**, 627–643 (2001).
11. Jickells, T. D. *et al.* Global iron connections between desert dust, ocean biogeochemistry, and climate. *Science* **308**, 67–71 (2005).
12. Sarmiento, J. L., Hughes, C. W., Stouffer, R. J. & Manabe, S. Simulated response of the ocean carbon cycle to anthropogenic climate warming. *Nature* **393**, 245–249 (1998).
13. Marinov, I., Gnanadesikan, A., Toggweiler, J. R. & Sarmiento, J. L. The Southern Ocean biogeochemical divide. *Nature* **441**, 964–967 (2006).
14. Coale, K. H. *et al.* Southern Ocean iron enrichment experiment: Carbon cycling in high- and low-Si waters. *Science* **304**, 408–414 (2004).
15. Boyd, P. W. *et al.* A mesoscale phytoplankton bloom in the polar Southern Ocean stimulated by iron fertilization. *Nature* **407**, 695–702 (2000).
16. Gervais, F., Riebesell, U. & Gorbunov, M. Y. Change in primary productivity and chlorophyll a response to iron fertilization in the Southern Polar Frontal Zone. *Limnol. Oceanogr.* **47**, 1324–1335 (2002).
17. Hart, T. J. Phytoplankton periodicity in Antarctic surface water. *Discov. Rep.* **VIII**, 1–268 (1942).
18. Sullivan, C. W., Arrigo, K. R., McClain, C. R., Comiso, J. C. & Firestone, J. Distribution of phytoplankton blooms in the Southern Ocean. *Science* **262**, 1832–1837 (1993).
19. Tyrell, T. *et al.* Effect of seafloor depth and phytoplankton blooms in high nitrate low chlorophyll (HNLC) regions. *J. Geophys. Res.* **110**, doi:10.1029/2005JG000041 (2005).
20. Blain, S. *et al.* A biogeochemical study of the island mass effect in the context of the iron hypothesis: Kerguelen Islands, Southern Ocean. *Deep-sea Res. I* **48**, 163–187 (2001).
21. Measures, C. I. & Vink, S. Dissolved Fe in the upper waters of the Pacific sector of the Southern Ocean. *Deep-sea Res. II* **48**, 3913–3941 (2001).
22. Timmermans, K. R. *et al.* Growth rates of large and small Southern Ocean diatoms in relation to availability of iron in natural seawater. *Limnol. Oceanogr.* **46**, 260–266 (2001).
23. Law, C., Abraham, E. R., Watson, A. & Liddicoat, M. Vertical eddy diffusion and nutrient supply to the surface mixed layer of the Antarctic Circumpolar Current. *J. Geophys. Res.* **108**, doi:10.1029/2002JC001604 (2003).
24. Boyd, P. W. *et al.* FeCycle: attempting an iron biogeochemical budget from a mesoscale SF₆ tracer experiment in unperturbed low iron waters. *Glob. Biochem. Cycles* **19**, doi:10.1029/2005GB002494 (2005).
25. Rutgers Van Der Loeff, M. M., Buesseler, K. O., Bathmann, U., Hense, I. & Andrews, J. Comparison of carbon and opal export rates between summer and spring bloom in the region of the Antarctic Polar Front, SE Atlantic. *Deep-sea Res. II* **49**, 3849–3869 (2002).
26. Buesseler, K. O., Andrews, J. E., Pike, S. M. & Charette, M. A. The effects of iron fertilization on carbon sequestration in the Southern Ocean. *Science* **304**, 414–417 (2004).
27. Bowie, A. R. *et al.* The fate of added iron during a mesoscale fertilisation experiment in the Southern Ocean. *Deep-sea Res. II* **48**, 2703–2743 (2001).
28. Gnanadesikan, A., Sarmiento, J. L. & Slater, R. D. Effects of patchy ocean fertilization on atmospheric carbon dioxide and biological production. *Glob. Biochem. Cycles* **17**, doi:10.1029/2002GB001940 (2003).
29. Buesseler, K. O. & Boyd, P. W. Will ocean fertilization work? *Science* **300**, 67–68 (2003).
30. Chisholm, S. W., Falkowski, P. G. & Cullen, J. J. Dis-crediting ocean fertilization. *Science* **294**, 309–310 (2001).
31. Wanninkhof, R. H. & McGillis, W. R. A cubic relationship between air-sea exchange and wind speed. *Geophys. Res. Lett.* **26**, 1889–1892 (1999).
32. Nightingale, P. *et al.* *In situ* elevation of the sea-air gas exchange parameterisations using novel conservative and volatile tracers. *Glob. Biochem. Cycles* **14**, 373–387 (2000).

Supplementary Information is linked to the online version of the paper at www.nature.com/nature.

Acknowledgements We thank the captain and the crew of the RV *Marion Dufresne*. This work was supported by the Institut National des Sciences de l'Univers (INSU) and the Centre National de la Recherche Scientifique (CNRS), l'Institut Paul Emile Victor (IPEV), French-Australian Science and Technology (FAST), the Australian Commonwealth Cooperative Research Centre programme through the Antarctic Climate and Ecosystem CRC, and Belgian Science Policy (BELSPO). The project benefited from collaboration with N. Metzl, leader of Ocean Indien Service d'Observation (OISO) supported by INSU, IPEV and Institut Pierre Simon Laplace (IPSL). We acknowledge the contributions of V. Barthaux, P. Catala, J. Caparros and J. Raz (technical assistance), and M.P. Jouandet (computation of the seasonal carbon budget).

Author Information Reprints and permissions information is available at www.nature.com/reprints. The authors declare no competing financial interests. Correspondence and requests for materials should be addressed to S.B. (stephane.blain@univmed.fr).

METHODS

Selection of stations and hydrographic sampling. Composite images of MODIS and MERIS processed by ACRI Co. and delivered daily to the RV *Marion Dufresne* were used to locate the stations inside and outside the bloom. The vessel seawater supply (intake at a depth of 5 m) was used to continuously monitor p_{CO_2} , DIC, temperature, salinity and fluorescence³³. Vertical sampling of the water column from the surface to the bottom was performed using 22 acid-cleaned 12-l Niskin bottles mounted on a CTD rosette system with a lowered acoustic Doppler current profiler (LADCP) and a C star transmissiometer (WETLABS). Clean trace metal samples were obtained using ten 12-l Go-Flo bottles mounted on a 2,500-m Kevlar line and triggered by Teflon messengers. *In situ* pumps were used to collect particles. The vertical diffusivity, K_z , was estimated from the vertical density profiles and from corresponding Thorpe scales^{34,35}.

Biogeochemical parameters. Dissolved iron³⁶ and organic ligands³⁷ present in filtered (<0.2 μm) sea water were measured on board in a clean laboratory. The accuracy of DFe determinations was checked using the new sub-nanomolar Fe standard sampled and analysed during the SAFE cruise³⁸. Samples from Go-Flo bottles were filtered (>0.2 μm) for particulate trace metal determinations³⁹. The crustal ratio PFe/PAl = 0.2 (ref. 40) was used to estimate PFe_{iih}. Discrete water samples taken from Niskin bottles were analysed for macronutrients⁴¹, DIC³³, DOC, POC⁴¹, and U/Th β -counting⁴². Rates of production were determined using 24-h deck incubations with simulated *in situ* light levels, using ¹³C (ref. 41). Uptake and regeneration rates of iron were determined using ⁵⁵Fe and a biogenic budget²⁷. The net iron demand was calculated as the difference between the total biogenic demand and supply. Gross community production and community respiration were concurrently determined by O₂ and DIC evolution using 24-h incubations with simulated light levels⁴³. ²³⁴Th export fluxes at 100 m and 200 m depth were estimated using a non-steady-state (NSS) one-dimensional model accounting for vertical diffusion (equations in ref. 44). ²³⁴Th fluxes at A3 were 2,999 and 5,217 d.p.m. m⁻² d⁻¹ at 100 and 200 m, respectively, and at C11, 2,003 and 2,281 d.p.m. m⁻² d⁻¹ at 100 and 200 m, respectively. Conversion to POC export was based on POC/²³⁴Th (C/Th) ratios of 5–210- μm particles from large volume filtrations at 100 m depth (7.7 and 6.1 $\mu\text{mol d.p.m.}^{-1}$ at A3 and C11, respectively). Estimates at 200 m depth were based on C/Th from 1–335- μm particles from free-drifting sediment trap collections at site A3 (4.7 $\mu\text{mol d.p.m.}^{-1}$) and on C/Th ratios from 5–210- μm particles from large volume filtrations at 130 m depth at C11 (4.5 $\mu\text{mol d.p.m.}^{-1}$).

33. Jabaud, A., Metz, N., Brunet, C., Poisson, A. & Schauer, B. Interannual variability of the carbon dioxide system in the Southern Indian Ocean (20°S–60°S): the impact of a warm anomaly in austral summer. *Glob. Biochem. Cycles* **18**, doi:10.1029/2002GB002017 (2004).
34. Osborn, T. R. Estimate of vertical diffusion from dissipation measurements. *J. Phys. Oceanogr.* **10**, 83–89 (1980).
35. Dillon, T. M. Vertical overturns: a comparison of Thorpe and Ozmidov length scales. *J. Geophys. Res.* **85**, 9601–9613 (1982).
36. Sarthou, G. *et al.* Atmospheric iron deposition and sea-surface dissolved iron concentrations in the eastern Atlantic Ocean. *Deep-Sea Res.* **150**, 1339–1352 (2003).
37. Croot, P. & Johansson, M. Determination of iron speciation by cathodic stripping voltammetry in seawater using the competing ligand 2-(2-thiazolylazo)-*p*-cresol (TAC). *Electroanalysis* **12**, 565–576 (2000).
38. Johnson, K. S. *et al.* Sampling and analysis of Fe: The SAFE iron intercomparison cruise. *Eos Trans. AGU* **87**(36), Ocean Sci. Meet. Suppl. (2006).
39. Cullen, J. T. & Sherrel, M. Techniques for determination of trace metal in small samples of size-fractionated particulate matter: phytoplankton metals off central California. *Mar. Chem.* **67**, 233–247 (1999).
40. Wedepohl, K. H. The composition of continental crust. *Geochim. Cosmochim. Acta* **59**, 1217–1232 (1995).
41. Fernandez, C., Raimbault, P., Caniaux, Y., Garcia, N. & Rimmelin, P. An estimation of annual new production and carbon budget in the northeast Atlantic Ocean during 2001. *J. Geophys. Res.* **110**, doi:10.1029/2004JC002621 (2005).
42. Pike, S. M., Buesseler, K. O., Andrews, J. & Savoye, N. Quantification of ²³⁴Th recovery in small volume of sea water samples by inductively coupled plasma-mass spectrometry. *J. Radioanal. Nucl. Chem.* **263**, 355–360 (2005).
43. Robinson, C. & Williams, P. J. I. Plankton net community production and dark respiration in the Arabian Sea during September 1994. *Deep-Sea Res.* **11** **46**, 745–765 (1999).
44. Savoye, N. *et al.* ²³⁴Th sorption and export models in the water column: a review. *Mar. Chem.* **100**, 234–249 (2006).